

## **SURFACTANT-POLYMER COMPOSITION FOR SUBSTANTIALLY SOLID-FREE WATER BASED DRILLING, DRILL-IN, AND COMPLETION FLUIDS**

### **5     Field of the Invention**

The present invention relates to water based drilling, drill-in, and completion fluids (hereafter collectively called "drilling fluids") comprising surfactant-polymer blends that are effective to generate low shear rate viscosity (LSRV), to provide fluid loss control, and to stabilize the dispersed phase. The water based drilling fluids  
10     preferably have a relatively low density and are substantially solid free.

### **Background of the Invention**

Drilling fluids serve various functions, such as promoting borehole stability, removing drilled cuttings from the wellbore, cooling and lubricating the bit and the drillstring, as well as controlling subsurface pressure. Specially formulated fluids are  
15     used in connection with completion and workover operations to minimize damage to the formation. Workover fluids are used during remedial work in the well, such as removing tubing, replacing a pump, logging, reperforating, and cleaning out sand or other deposits.

Formation damage can result when solids and/or filtrate derived from one of  
20     these fluids invades the formation during drilling operations. Such formation damage may be minimized by treating the well in a near-balanced condition (wellbore pressure close to formation pressure). Unfortunately, in high-pressure wells, the well often must be treated in an over-balanced and/or an under-balanced condition. If the well is over-balanced, the treating fluid is designed to temporarily seal perforations to  
25     prevent entry of fluids and solids into the formation. If the well is under-balanced, the

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treating fluid is designed to prevent entry of solids from the formation into the wellbore.

Current under-balanced drilling technologies use air, foams and aerated fluids. These air-based technologies solve some of the problems encountered drilling in under-balanced conditions; however, air based technologies are very costly.

Low density fluids have the capacity to control over-balanced and under-balanced formations. Water-based low density fluids are of particular interest because of environmental superiority and relatively low cost. In order to be effective, low density drilling fluids must have adequate rheology and fluid loss control properties. Viscosity and fluid loss may be controlled in low density fluids by adding certain polymers to the fluid. Solid bridging agents also generally are required to prevent fluid loss.

Unfortunately, solid bridging agents may plug pore throats in the reservoir rock. It would be desirable to formulate fluids which have the necessary rheology and fluid loss control properties without the need to use solid bridging agents.

### **Summary of the Invention**

The invention provides water-based drilling fluids having effective rheology with low shear rate viscosity and effective fluid control properties comprising water soluble polymers and surfactant adapted to associate with said water soluble polymers and to provide said effective rheology, and effective fluid loss control properties.

### **Detailed Description of the Invention**

The present invention provides water based drilling fluids comprising surfactant-polymer blends that are effective to generate low shear rate viscosity (LSRV), to provide effective fluid loss control, and to stabilize the dispersed phase in

the water based fluid. The water based drilling fluids preferably have relatively low density and are substantially solid free.

The drilling fluid preferably comprises water as a continuous phase and an internal phase comprising a water emulsifiable material. The water emulsifiable material comprises a concentration sufficient to provide effective lubrication. A preferred concentration of the water emulsifiable material is from about 2 to about 20 vol.%, most preferably about 5 vol.%.

The water emulsifiable material preferably is non-toxic. As used herein, the term "non-toxic" is defined to mean that a material meets the applicable EPA requirements for discharge into U.S. waters. Currently, a drilling fluid must have an LC<sub>50</sub> (lethal concentration where 50% of the organisms are killed) of 30,000 parts per million (ppm) suspended particulate phase (SPP) or higher to meet the EPA standards.

Suitable water-emulsifiable materials are water-insoluble and include, but are not necessarily limited to olefins, paraffins, water insoluble polyglycols, water insoluble esters, water insoluble Fischer-Tropsch reaction products, and other non-toxic organic materials, and combinations thereof. Suitable olefins are branched and/or linear and preferably are relatively non-toxic synthetic olefins. Examples of preferred olefins include but are not necessarily limited to esters, polialphaolefins, linear alpha olefins, and internal olefins, typically skeletally isomerized olefins. Most preferred olefins are described in U.S. Patent Nos. 5,605,872 and 5,851,958, incorporated herein by reference. Preferred paraffins are described in U.S. Patent No. 5,837,655, incorporated herein by reference. Suitable polyglycols have a molecular weight sufficiently high to render them water insoluble. Preferred polyglycols are water insoluble polypropylene glycols. A most preferred polyglycol has an average

molecular weight of at least about 2000, more preferably at least about 4000. Suitable polyglycols are available from a variety of commercial sources. A preferred polyglycol is POLYGLYCOL P-4000, available from Dow Chemical Co.

The drilling fluid also comprises water-soluble polymers or copolymers, hereinafter collectively referred to as "water-soluble polymers," that are capable of viscosifying the drilling fluid and/or providing fluid loss control for the drilling fluid. Preferred polymers are non-toxic and include, but are not necessarily limited to water soluble starches and modified versions thereof, water soluble polysaccharides and modified versions thereof, water soluble celluloses and modified versions thereof, and water soluble polyacrylamides and copolymers thereof. Generally, the quantity of polymer used is at least about 2 lb/bbl. or more, preferably about 7.5 lb/bbl. or more.

Starches that are suitable for use in the drilling fluid include, but are not necessarily limited to corn based and potato based starches, preferred starches being more temperature stable starches. Polysaccharides that are suitable for use in the present invention include, but are not necessarily limited to xanthan polysaccharides, wellan polysaccharides, scleroglucan polysaccharides, and guar polysaccharides. Celluloses that are suitable for use in the present invention include, but are not necessarily limited to hydrophobically modified hydroxyethyl celluloses and cationic cellulose ethers. Suitable copolymers of acrylamide include copolymers with acrylate monomers, hydrophobic N-isopropylacrylamide, and the like.

As used herein, the terms "modified starches" and "modified polysaccharides" or "synthetically modified polysaccharides" refer to starches and polysaccharides that have been chemically modified in a manner that renders them inherently non-fermentable in order to avoid the need for a preservative. Water-

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soluble “modified starches” and “modified polysaccharides” that should operate successfully as water-soluble polymers include, but are not necessarily limited to: hydroxyalkyl starches and polysaccharides; starch and polysaccharide esters; cross-link starches and polysaccharides; hypochlorite oxidized starches and

- 5 polysaccharides; starch and polysaccharide phosphate monoesters; cationic starches and polysaccharides; starch and polysaccharide xanthates; and, dialdehyde starches and polysaccharides. These derivatized starches and polysaccharides can be manufactured using known means, such as those set forth in detail in Chapter X of *Starch: Chemistry and Technology* 311-388 (Roy L. Whistler, et al. eds., 1984),
- 10 incorporated herein by reference.

- Specific examples of suitable modified starches and modified polysaccharides include, but are not necessarily limited to: carboxymethyl starches and polysaccharides; hydroxyethyl starches and polysaccharides; hydroxypropyl starches and polysaccharides; hydroxybutyl starches and polysaccharides;
- 15 carboxymethylhydroxyethyl starches and polysaccharides; carboxymethylhydroxypropyl starches and polysaccharides; carboxymethylhydroxybutyl starches and polysaccharides; epichlorohydrin starches and polysaccharides; alkylene glycol modified starches and polysaccharides; and, other starch and polysaccharide copolymers having similar characteristics. Preferred
- 20 modified starches and/or modified polysaccharides comprise a functional group selected from the group consisting of a carboxymethyl group, a propylene glycol group, and an epichlorohydrin group.

A preferred polymer is a blend of modified polysaccharides and synthetically modified starch, preferably a blend comprising from about 10 wt.% to about 90 wt.%

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modified polysaccharide and from about 10 wt.% to about 90 wt.% synthetically modified starch. More preferably, the blend is from about 40 to about 60 wt.% synthetically modified polysaccharide with the remainder of the blend being synthetically modified starch. In a most preferred embodiment, the blend is a 50/50 wt.% blend of modified polysaccharide/synthetically modified starch.

Suitable modified polysaccharides have a weight average molecular weight of about 500,000 to about 2,500,000, preferably from about 700,000 to about 1,200,000, most preferably about 1,000,000. Preferred modified polysaccharides for use in the blend are xanthan polysaccharides. Suitable modified xanthan polysaccharides are commercially available and include, but are not necessarily limited to XAN-PLEX™, XAN-PLEX D™, XANVIS™, all of which are available from Baker Hughes INTEQ. A most preferred modified polysaccharide is commercially available as XAN- PLEX™ D from Baker Hughes INTEQ.

Suitable synthetically modified starches have a weight average molecular weight of from about 200,000 to about 2,500,000, preferably from about 600,000 to about 1,000,000. Preferred synthetically modified starches comprise a functional group selected from the group consisting of a carboxymethyl group, a propylene glycol group, and an epichlorohydrin group. Suitable synthetically modified starches include, but are not necessarily limited to BIOPAQ™, BIOLOSE™, and PERMALOSE™, available from Baker Hughes INTEQ. A most preferred commercially available product is BIOLOSE™, available from Baker Hughes INTEQ. A preferred blend is 50/50 wt.% XAN-PLEX™ D and BIOLOSE™.

The drilling fluid comprises a quantity of one or more surfactants which bind to the polymer. The surfactant(s) are believed to associate with the polymers,

lowering interfacial properties, such as surface tension, and forming complex associations between the polymer and surfactant. The result is increased viscosification or gelation characterized by a micellar-type aggregate along the backbone of the polymer chain.

5       The type of surfactant used in a given fluid will vary with the type of polymer and the charge of pendant groups on the polymer. Where the polymer comprises groups susceptible to hydrogen bonding, the surfactant is substantially non-ionic, and more susceptible to forming hydrogen bonds with the surfactant. Where the polymer comprises pendant groups that are susceptible to ionic bonding, the surfactant is  
10    cationic or anionic. A variety of surfactants may be used as long as the surfactant bonds or associates with pendant groups on the polymer to produce three dimensional structures that are effective to increase the viscosity of the drilling fluid to a desired level.

      The viscosity of a fluid is its internal resistance to flow as measured in  
15   centipoise units. The coefficient of viscosity of a normal homogeneous fluid at a given temperature and pressure is a constant for that fluid and independent of the rate of shear or the velocity gradient. Fluids that obey this rule are "Newtonian" fluids. In fluids called "non-Newtonian fluids," this coefficient is not constant but is a function of the rate at which the fluid is sheared as well as of the relative concentration of the  
20   phases. Drilling fluids generally are non-Newtonian fluids. Non-Newtonian fluids frequently exhibit plastic flow, in which the flowing behavior of the material occurs after the applied stress reaches a critical value or yield point (YP). Yield points in drilling fluids are frequently expressed in units of pounds per square 100 feet, wherein the yield point is a function of the internal structure of the fluid.

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In drilling, once the critical value or yield point (YP) of the drilling fluid is achieved, the rate of flow or rate of shear typically increases with an increase in pressure, causing flow or shearing stress. The rate of flow change, known as plastic viscosity (PV), is analogous to viscosity in Newtonian fluids and is similarly measured in centipoise units. In drilling fluids, yield points (YP) above a minimum value are desirable to adequately suspend solids, such as weighting agents and cuttings. The fluid preferably has a yield point of from about 20 to about 50, preferably 30 or more pounds per 100 square feet.

The polymer and surfactant produce a drilling fluid having "low shear rate viscosity," defined herein as a viscosity of at least about 70,000 upon exposure to 0.3 rpm, measured with a Brookfield viscometer at 75 °F. Preferably, the polymer/surfactant structures produce a viscosity of at least about 100,000 cP, preferably about 200,000 cP upon exposure to 0.3 rpm.

The surfactant preferably aids in emulsifying the water emulsifiable material, producing droplets having an average diameter as small as possible. The droplet size will depend upon the water emulsifiable material used, and may be relatively large as long as the resulting fluid has effective rheology and fluid loss control properties. Suitable average diameters for the droplets are about 35 microns or less, preferably about 20 microns or less, more preferably about 15 microns or less, and most preferably about 5 microns or less. The surfactant also reduces the surface tension of the water based fluid. The surface tension of water is about 74 nN/m. Preferred surfactants reduce the surface tension of the water based fluid to from about 25 to about 40 nN/m, preferably to about 36 nN/m.



Suitable surfactants include, but are not necessarily limited to alkyl sulfates, alkyl ether sulfates, alkyl sulfonates, ethoxylated esters, ethoxylated glycoside esters, alcohol ethers, and phosphated esters comprising about 8 to about 18 carbon atoms, preferably about 8 to about 12 carbon atoms, and salts thereof. Preferred surfactants are selected from the group consisting of alkyl sulfates and alkyl ether sulfates, more preferably salts of a trialkyl ether sulfates, most preferably sodium tridecyl ether sulfate. The quantity of surfactant used is sufficient to achieve the foregoing rheological properties. The amount of surfactant used typically is from about 0.2 lb./bbl. to about 4 lb./bbl., preferably about 2 lb./bbl.

Where a fluid is expected to encounter relatively high temperatures, typically about 220 °F. or more, the fluid preferably comprises a salt effective to increase thermal stability. The salt preferably comprises a compound selected from the group consisting of a sulfate and a sulfonate. The salt preferably is an alkali metal salt of a compound selected from the group consisting of a sulfate and a sulfonate, most preferably an alkali metal salt of a compound selected from the group consisting of a thiosulfate and a thiosulfonate. A most preferred salt is sodium thiosulfate. Where necessary, the salt preferably comprises from about 0.2 lb./bbl. to about 3.0 lb./bbl., preferably about 1.5 lb./bbl.

It is preferred for the drilling fluid to be substantially solids free, however, other conventional additives may be used in the fluid. Such additives include, but are not necessarily limited to shale stabilizer(s), filtration control additive(s), suspending agent(s), dispersant(s), thinner(s), anti-balling additive(s), lubricant(s), weighting agent(s), seepage control additive(s), lost circulation additive(s), drilling enhancer(s), penetration rate enhancer(s), corrosion inhibitor(s), acid(s), base(s), buffer(s),

scavenger(s), gelling agent(s), cross-linker(s), catalyst(s), soluble salts, biocides, and combinations thereof. Although it is preferred for the drilling fluid not to contain solid bridging agents, one or more bridging and/or weighting agents may be added to the fluid. Suitable shale stabilizers include, but are not necessarily limited to

5 polyglycols, inorganic salts, chelates, amines, alkanolamines, alkanolamides, amphoteric compounds, alone or in aqueous solutions, and mixtures thereof. Suitable shale stabilizing inorganic salts include, but are not necessarily limited to alkali metal salts, silicate salts, and aluminum salts. Acids include acids used to treat cement contamination.

10 The fluid is prepared using conventional procedures, with the polymer and surfactant being added before the dispersant and water emulsifiable material, if used. The pH of the fluid is measured and, if needed, adjusted to from about 8.5 to about 11.5, preferably about 9.5. The pH preferably is adjusted using a suitable organic base as a buffer. Substantially any buffer may be used as long as it does not interfere

15 with the surfactant and does not affect the stability of the generated droplet size. Preferred buffers include, but are not necessarily limited to ethanolamines, alkali metal hydroxides, preferably sodium or potassium hydroxide, alkali metal acetates, preferably sodium or potassium acetate. Preferred buffers are ethanolamines, more preferably mono-ethanolamine or diethanolamine, and most preferably mono-

20 ethanolamine.

The drilling fluids have a relatively low density depending upon the particular quantities and components used. The density generally is about 7.9 lb./bbl. or more, preferably about 8.1 lb./bbl. or more. The drilling fluids exhibit low shear rate viscosity (LSRV), which resists fluid movement into the formation zone and inhibits

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lost circulation. The fluids comprise a stabilized emulsion of very small droplets of water emulsifiable material stabilized by the surfactant polymer complex. The stabilized droplets help to control filtrate invasion. The fluids exhibit high viscosity at low shear rate, high yield point, low plastic viscosity, high gel strength, with thermal stability up to 300°F. Excellent fluid loss control is obtained at differential pressures of about 50 psi or more, preferably up to about 1500 psi or more. The term "effective fluid loss control properties" is defined to mean that, using the "standard dynamic filtration fluid loss test" (API Recommended Practice 13B-1), the measured fluid loss is about 18 ml./30 min. or less. Preferably the fluid loss is 5 ml./30 min. or less, more preferably about 1 ml./30 min. or less, even more preferably about 0.5 ml./30 min. or less. The fluids exhibit a lower surface tension than most water based drilling fluids, which maintain or improve the mobility of the crude oil in water-wet reservoirs, leading to better productivity. The absence of solid bridging agents in preferred formulations decreases the risk of plugging of the pore throats in the reservoir rock.

The drilling fluids also prevent fluid losses to the formation in depleted reservoirs.

The invention will be better understood with reference to the following examples, which are illustrative only, and should not be construed as limiting the invention:

#### **Selection of surfactants**

Surfactant screening was performed using various anionic and nonionic surfactants, including branched alcohol ethoxylates, sodium alkyl sulfates, and sodium ether sulfates.

Surfactant screening for the isomerized olefin internal phase in-water and high molecular weight polyglycol-in-water emulsification shows that the most effective

surfactant is sodium tridecyl ether sulfate. This surfactant generated emulsions with initial droplet size of less than 10 microns, which remained stable after hot rolling at 200°F. However, the droplet size increased considerably (78 microns) when the emulsions were hot rolled at 250 °F. The increased droplet size could be overcome by using a higher concentration of surfactant or by mixing the surfactant with other additives, such as a polymer, that could improve thermal stability of the drilling fluid.

***Screening of surfactants in a 70/30 olefin base oil/water emulsification***

Surfactant	Droplet Size, microns	
	Initial	After 250°F Hot Roll
Isotridecyl alcohol ethoxylate, 10 EO	7.8	Unstable O/W
Oleyl alcohol ethoxylate, 20 EO	8.5	Unstable O/W
Sodium octyl sulfate	68.0	Unstable O/W
Sodium linear alcohol ether (6) phosphate	14.2	Unstable O/W
Sodium tridecyl ether sulfate	7.2	78.7

Evaluation and selection of polymers was performed in conjunction with surfactant. Testing at temperatures up to 250 °F was carried out with a combination of surfactant and polymers, in order to define the best product combination and optimum concentrations to produce a fluid with low shear rate viscosity, high yield point, low plastic viscosity, and high gel strength. Commercially available polysaccharides and starches were used for viscosity and filtration control. Product selection was based on evaluation of rheology and emulsion droplet size after fluids were hot rolled at 200°F and 250°F for 16 hours. The polymers used in the following experiments were selected. The polymers were used at a 50:50 ratio, unless otherwise indicated.

### **Example 1**

A polymer blend (XANTHAN D/BIOLOSE) with the surfactant (sodium ether sulfate) was used in a base formulation with 95/5 water/isomerized olefin ratio, 6 lb./bbl. polymer blend, 0.5 lb./bbl. monoethanolamine (MEA), and 5 lb./bbl. sodium formate, and between 0.5 and 6 lb./bbl. of surfactant.

The fluids were mixed in a baffled two-liter conical mixing cup, using a Prince Castle mixer operating initially at about 8500 rpm to mix the polymer in water; then the mixing speed was reduced between 6000 to 6500 rpm to reduce foam formation. A defoamer may be used, if necessary. The order of mixing is not critical. The following Table describes the order of addition in this experiment. The non-polar internal phase was isomerized olefins, specifically GT-2700<sup>TM</sup> obtained from Baker Hughes INTEQ), polypropylene glycol P-2000 (Dow Chemical Company), or P-4000 (Dow Chemical Company).

#### ***Additives by order of addition***

<b>Additives</b>
Water
Viscosifier + Filtration Control
Buffer
Surfactant
Dispersed or internal phase (oil or glycol)
Shale inhibitor and others

The following Table and Figure 1 give the droplet size reduction with increasing surfactant concentration in the described base formulation.

*Effect of surfactant concentration on mean droplet size of base formulations*

Surfactant Concentration, lb/bbl	Internal Phase in Water	Droplet Size, microns	
		Initial	After 250°F Hot Roll
0.5	Olefin	7.8	9.4
3.0	Olefin	4.4	4.2
6.0	Olefin	3.0	3.5
0.5	Polyglycol P-400	12.5	33.3
6.0	Polyglycol P-400	7.2	15.5

The foregoing demonstrates a reduction from 9.4 to 3.5 microns by changing surfactant concentration from 0.2 to 2 lb./bbl. in formulations with isomerized olefin.

- 5 This effect was more noticeable in polyglycol formulations that exhibited decreased droplet size from 33.3 to 15.5 microns as surfactant concentration was increased from 0.2 to 2 lb/bbl. The olefin droplets form part of the viscoelastic complexes that control the formation and, at the same time, function to give some lubricity to the fluid.

- 10 MEA was used for pH control because it did not interfere with the surfactant, and therefore did not affect the stability of the generated droplet size. Magnesium oxide was evaluated also as a pH buffer, but is not preferred because of slightly increased droplet size. Surfactant screening was performed using various anionic and nonionic surfactants, including branched alcohol ethoxylates, sodium alkyl sulfates, and sodium alkyl ether sulfates.
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**Example 2*****Effect of temperature in the formulations***

To obtain the desired rheological properties and also to maintain thermal stability up to 270 °F, other additives were evaluated in conjunction with the base formulation. Sodium thiosulfate was found to improve the thermal stability of the drill-in fluid formulations. The following Table show the formulations used:

***Formulations for temperature evaluation***

Additives	Formulation for application < 200 °F	Formulation for application up to 270 °F
Water, bbl	0.95	0.95
Isomerized olefin bbl	0.05	0.05
Polymer blend, lb/bbl	6	6
Monoethanolamine, lb/bbl	0.5	0.5
Surfactant, lb/bbl	1	1
Sodium formate, lb/bbl	5	5
Sodium thiosulfate, lb/bbl	---	1.0

The following rheological properties were obtained with and without sodium thiosulfate:

10 ***Rheological properties (without sodium thiosulfate, < 200 °F)***

Properties	Measured at 120 °F using FANN 35 Viscometer				
	Initial	After H.R. at 150 °F	After H.R. at 200 °F	After H.R. at 220 °F	After H.R. at 250 °F
6 rpm	25	27	28	17	16
3 rpm	22	24	25	13	12
10-sec Gel	23	25	26	13	12
10-min Gel	28	32	33	15	15
30-min Gel	31	33	35	16	16
PV, cP	11	10	11	10	9
YP, lbf/100 ft <sup>2</sup>	45	48	48	43	43

H.R.: Hot-rolled for 16 hours

***Rheological properties (with sodium thiosulfate, up to 270 °F)***

Properties	Measured at 120 °F using FANN 35 Viscometer					
	Initial	After H.R. at 150 °F	After H.R. at 200 °F	After H.R. at 220 °F	After H.R. at 250 °F	After H.R. at 270 °F
6 rpm	25	27	28	28	25	25
3 rpm	22	24	25	25	22	22
10-sec Gel	23	25	26	26	22	22
10-min Gel	28	32	33	29	25	25
30-min Gel	31	33	35	30	26	27
PV, cP	11	10	11	8	8	8
YP, lbf/100 ft <sup>2</sup>	45	48	48	47	48	47

H.R.: Hot-rolled for 16 hours

- Formulations without sodium thiosulfate had significantly decreased low rpm readings, as well as decreased gel strength, when the sample was hot rolled at temperatures higher than 200 °F. Addition of 0.5 lb./bbl. of sodium thiosulfate in formulations hot-rolled at 250 °F and 1 lb./bbl. of sodium thiosulfate in formulation hot-rolled at 270 °F maintained the rheological properties of the fluids.

**Example 3*****10 Rheological properties and dynamic filtration evaluation***

The following shows polyglycol/water formulations with various concentrations of surfactant and the properties of these formulations, measured after hot rolling at 200°F:



*Effect of surfactant concentration on mud formulation containing polyglycol as a dispersed phase*

Additives	Formulation 1	Formulation 2	Formulation 3
Water, bbl	0.95	0.95	0.95
Modified polysaccharide polymer/ starch derivative, lb/bbl	6	6	6
Surfactant, lb/bbl	0.5	3	6
Monoethanolamine, lb/bbl	0.5	0.5	0.5
Sodium formate, lb/bbl	5	5	5
POLYGLYCOL P-2000, bbl	0.05	0.05	0.05
REV-DUST, lb/bbl	---	---	---
120°F Properties After Hot Roll at 200°F	Formulation 1	Formulation 2	Formulation 3
6 rpm	29	29	28
3 rpm	26	26	25
Plastic viscosity, cP	10	11	11
Yield point, lb/100 ft <sup>2</sup>	50	50	47
10 s/10 min/30 min gels, lb/100 ft <sup>2</sup>	27/33/34	27/32/34	25/32/33
Density, lb/gal	7.9	7.7	7.8
PH	9.8	9.7	9.7
Mean particle size, microns (without Rev-dust)	33	18	15.5
( <sup>1</sup> )Spurt loss ml/30 min filtrate*2	6/15.2	5/13	2.8/10.6

(1) 2 % REV-DUST™, a simulated drilled product which may be obtained from Mil-White Company, Houston, Texas, was added to the samples used for the dynamic filtration test.

Rheological properties did not significantly change with increasing surfactant concentration from 0.2 to 2 lb/bbl. The results also indicate that increased concentration of surfactant decreased spurt loss and the total accumulated filtrate volume measured in the dynamic filtration test.

Without being limited to a particular theory or mechanism of action, the surfactants are believed to have an attractive interaction with modified polysaccharide polymers (cycloliner or branched) and macromolecular starch derivatives, and to form complexes sometimes described as a "Necklace model." This structure is

believed to be characterized by a micellar-type aggregate along the backbone of the polymer chain that could be responsible for the reduction of filtrate volume in the evaluated formulations.

#### **Example 4**

- 5           Formulations with isomerized olefins as the internal phase were evaluated by varying the additive concentration with the following results:

#### ***Mud formulation containing 7.5 lb/bbl of polymers blend***

<b>Additives</b>	<b>Base Formulation</b>	<b>120°F Properties after hot roll at 200°F</b>	
Water, bbl	0.95	6-rpm reading	28
Modified polysaccharide polymer/ starch derivative, lb/bbl	7.5	3-rpm reading	23
Surfactant, lb/bbl	6	Plastic viscosity, cP	19
Monoethanolamine, lb/bbl	0.5	Yield point, lb/100 ft <sup>2</sup>	60
Sodium formate, lb/bbl	5	10-sec gels, lb/100 ft <sup>2</sup>	26
Isomerized olefins, bbl	0.05	10-min gels, lb/100 ft <sup>2</sup>	30
		30-min gels, lb/100 ft <sup>2</sup>	32
		pH	9.8
		Droplet size, microns	3.5

- The foregoing Table and Fig. 2 indicates that a formulation containing 7.5 lb/bbl of polymers blend and 2 lb/bbl of surfactant produces a fluid with acceptable rheological properties.

- The rheological properties measured with Fann 35 exhibit (1) high initial gel strengths without evidence of progressively increasing gels, (2) high values for low rpm reading, and (3) relatively low plastic viscosities
- 15   • The average droplet sizes obtained using isomerized olefins as the internal phase were smaller than those obtained with the formulations containing P-2000 as internal phase (3 microns vs 15 microns).

This formulation was also evaluated using ECO-FLOW™ (paraffin oil) instead of isomerized olefins, and the properties of the formulated fluids were very similar.

### **Example 5**

Low shear rate viscosity (LSRV) was measured using the Brookfield viscometer at 75 °F, with the following results:

#### ***Low shear rate viscosity (LSRV) measured with the Brookfield viscometer***

<b>Additives</b>	<b>Formulation 1</b>	<b>Formulation 2</b>	<b>Formulation 1 + 2% Rev-Dust</b>	<b>Formulation 2 + 2% Rev-Dust</b>
Water, bbl	0.95	0.95	0.95	0.95
Modified polysaccharide polymer/starch derivative, lb/bbl	6	7	6	7.5
Surfactant, lb/bbl	6	6	6	6
MEA, lb/bbl	0.5	0.5	0.5	0.5
Sodium formate, lb/bbl	5	5	5	5
Isomerized olefins, bbl	0.05	0.05	0.05	0.05
Rev-Dust, % wt/wt	---	---	2	2
<b>Dial Reading After Hot Roll at 200°F</b>	<b>Brookfield Viscosity, cP at 75±1 °F</b>			
60 rpm	1224	2949	1839	4229
30 rpm	1996	5019	2987	7278
12 rpm	3899	10423	5939	15230
6 rpm	6589	18296	10138	27027
3 rpm	11198	32493	17463	47990
1.5 rpm	19196	55855	30553	85982
0.6 rpm	39659	122474	64853	181961
0.3 rpm	68652	196958	115175	318932

The results show a thixotropic behavior, with viscosity on the order of 70,000 to 200,000 cP at 0.3 rpm, dropping to a range of 1220 to 1839 at 60 rpm

**Example 6*****Dynamic filtration of formulations***

Dynamic filtration of formulations, without solid bridging agents, was measured  
 5 using 3 microns & 0.4 D aloxite disks, with the following results:

***Dynamic filtration.***

Additives	Formulation 1, Cumulative Filtration, ml T- 200°F Speed = 100 rpm Disk with 0.4 D and 3 microns
Water, bbl	0.95
Modified polysaccharide polymer/ starch derivative, lb/bbl	6
Surfactant, lb/bbl	6
Monoethanolamine, lb/bbl	0.5
Sodium formate, lb/bbl	5
GT-2700™, bbl	0.05
Rev-Dust, % wt/wt	2
Time, minutes	ΔP= 500 psi
0	0
1	5.0
5	6.4
10	6.6
15	7.2
20	7.8
25	8.0
30	8.4
Spurt loss/ 30 min filtrate*2, ml	5.0/ 16.2

The foregoing results indicate that formulations contaminated with 2% of solid  
 produced a spurt loss of 5 mL and filtrate volumes of 16.2 mL.

**Example 7*****Surface tension evaluation***

Surface tension strongly influences the flow of the fluid by reducing the surface viscosity and the surface elasticity. Surface tension of solutions containing

5 (1) polymer blend (2) polymer blend and surfactant were measured using the Wilhelmy plate method. The surface tension of water was measured as a reference. The following were the results:

***Measurement of surface tension***

Sample	Concentration (g/L)	Surface Tension (mN/m)
Water	--	71
Polymer / Water	2	66
Surfactant / Polymer / Water	0.1 (surfactant) 2 (polymers blend)	54.8
Surfactant / Polymer / Water	0.25 (surfactant) 2 (polymers blend)	41.6
Surfactant / Polymer / Water	0.5 (surfactant) 2 (polymers blend)	35.4

10 The foregoing results indicate:

- Little surface activity was observed with the sample of polymer blend in water, which only reduced the surface tension of the water from 71 to 66 mN/m.
  - The addition of 0.5 g/L of surfactant to the polymer solution generated a
- 15 reduction of surface tension from 66 to 35.4 nN/m.

Persons of ordinary skill in the art will recognize that many modifications may be made to the present invention without departing from the spirit and scope of the present invention. The embodiment described herein is meant to be illustrative only and should not be taken as limiting the invention, which is defined in the following

5 claims.

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